



Interpretation of zircon coronae textures from metapelitic granulites of the Ivrea–Verbano Zone, northern Italy: two-stage decomposition of Fe–Ti oxides

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Abstract. In this study, we report the occurrence of zircon coronae textures in metapelitic granulites of the Ivrea–Verbano Zone. Unusual zircon textures are spatially associated with Fe–Ti oxides and occur as (1) vermicular-shaped aggregates 50–200 μm long and 5–20 μm thick and as (2) zircon coronae and fine-grained chains, hundreds of micrometers long and $\leq 1 \mu\text{m}$ thick, spatially associated with the larger zircon grains. Formation of such textures is a result of zircon precipitation during cooling after peak metamorphic conditions, which involved: (1) decomposition of Zr-rich ilmenite to Zr-bearing rutile, and formation of the vermicular-shaped zircon during retrograde metamorphism and hydration; and (2) recrystallization of Zr-bearing rutile to Zr-depleted rutile intergrown with quartz, and precipitation of the submicron-thick zircon coronae during further exhumation and cooling. We also observed hat-shaped grains that are composed of preexisting zircon overgrown by zircon coronae during stage (2). Formation of vermicular zircon (1) preceded ductile and brittle deformation of the host rock, as vermicular zircon is found both plastically and cataclastically deformed. Formation of thin zircon coronae (2) was coeval with, or immediately after, brittle deformation as coronae are found to fill fractures in the host rock. The latter is evidence of local, fluid-aided mobility of Zr. This study demonstrates that metamorphic zircon can nucleate and grow as a result of hydration reactions and mineral breakdown during cooling after granulite-facies metamorphism. Zircon coronae tex-

tures indicate metamorphic reactions in the host rock and establish the direction of the reaction front.

1 Introduction

1.1 Growth of metamorphic zircon

Growth of zircon during metamorphism is described in a variety of lithologies and can occur at different metamorphic grades, from high- (e.g., Fraser et al., 1997, 2004; Degeling et al., 2001; Möller et al., 2003; Wu et al., 2006; Harley et al., 2007; Zhao et al., 2015) to low-temperature metamorphism, including low-temperature hydrothermal reactions (e.g., Dempster et al., 2004, 2008; Rasmussen, 2005; Hay and Dempster, 2009; Hay et al., 2010; Kohn et al., 2015). Growth and new precipitation of zircon has been traced under temperatures as low as 250 °C (Rasmussen, 2005). Metamorphic zircon can (a) precipitate from fluid or melt (e.g., Rasmussen, 2005; Kohn et al., 2015), (b) can result from breakdown of Zr-bearing phases (major or accessory) (e.g., Davidson and van Breemen, 1988; Fraser et al., 1997, 2004; Degeling et al., 2001), or (c) can exsolve from Zr-bearing phases (e.g., Bingen et al., 2001; Tomkins et al., 2007).

- a. New metamorphic zircon precipitates during exhumation and cooling of the host rock (Rasmussen, 2005; Kohn et al., 2015) as a result of partial or complete

dissolution of preexisting zircon (e.g., Dempster et al., 2008) in partial melt or metamorphic fluid during high-temperature metamorphism (Ewing et al., 2014). Solubility of zircon in most natural fluids is very low (Tromans, 2006); thus, zircon dissolution models mostly describe interactions with the melt (e.g., Harrison and Watson, 1983; Watson and Harrison, 1983). However, under high (prograde) temperatures and in fluids of favorable composition, zircon can be dissolved without melt involvement (e.g., Rubatto et al., 2008; Hay and Dempster, 2009; Hay et al., 2010; Ewing et al., 2014; Kohn et al., 2015).

- b. Zircon growth and overgrowth formation during cooling stage and/or retrograde metamorphism may also result from metamorphic reactions and breakdown of other Zr-bearing minerals (Fraser et al., 1997, 2004; Degeling et al., 2001; Möller et al., 2002, 2003; Tomkins et al., 2007). Fraser et al. (1997) and Möller et al. (2002) suggested that the source of newly precipitated zircon is Zr-bearing rock-forming phases (e.g., garnet), which experience breakdown and release Zr. The released Zr is not compatible with the breakdown product (e.g., with cordierite) and thus has to form a separate Zr phase, which could be zircon (Degeling et al., 2001; Möller et al., 2003). Zircon precipitation from other phases may also be facilitated by fluid. For example, Fraser et al. (2004) documented zircon rims precipitated during cooling stage from the hydrous fluid phase, which originated locally as a result of chlorite breakdown. The reactions with zircon precipitation in metamorphic rocks may be more efficient in the zones available for fluid infiltration, like fractures and shear zones (Bingen et al., 2001).
- c. Zircon exsolution has been observed in nature (e.g., Ewing et al., 2013; Pape et al., 2016) and has been demonstrated experimentally with Zr-rich rutile (Tomkins et al., 2007). Resulting zircon appears as thin exsolution lamellae or as small individual euhedral grains within rutile. Similarly, the metapelites from the Ivrea-Verbano Zone (IVZ) reveal thin zircon needles in rutile and chains of fine zircon grains framing rutile (Ewing et al., 2013; Pape et al., 2016).

In this contribution, we investigate unusual zircon textures, such as coronae found in dehydrated metapelitic granulites of the IVZ. We start with a review of the process of zircon precipitation from various Zr-bearing phases, followed by an overview of known examples of zircon coronae. After a short geological background of the unit, we describe the sampled outcrop as well as the sample itself macroscopically. Then a short exposition of applied methods and microscopic description of the studied sample are presented, followed by a detailed depiction of observed zircon microstructures and textures. For the sake of completion, we also include micro-

probe data of the studied sample. In the discussion, we suggest mineral reactions that could result in the formation of observed zircon coronae textures and then discuss the implications of our findings.

1.2 Zr-bearing phases potentially associated with zircon precipitation

Zircon dissolution and growth during metamorphism are not independent processes but must be coupled with the breakdown–growth of other phases and/or with various mineral–fluid reactions in the host rock (e.g., Tomkins et al., 2007; Austrheim et al., 2008). Metamorphic precipitation of zircon could be a result of the breakdown of and/or exsolution from various Zr-bearing phases (Davidson and van Breemen, 1988) such as garnet, amphibole, (clino)pyroxene and ilmenite (e.g., Fraser et al., 1997; Degeling et al., 2001; Möller et al., 2003; Söderlund et al., 2004; Harley et al., 2007; Kelsey et al., 2008; Morisset and Scoates, 2008), hemo-ilmenite (Morisset et al., 2005), baddeleyite (Bingen et al., 2001; Söderlund et al., 2004), rutile (Harley et al., 2007; Tomkins et al., 2007; Morisset and Scoates, 2008; Kelsey and Powell, 2011; Ewing et al., 2013, 2014; Pape et al., 2016), epidote, titanite (Kohn et al., 2015), chlorite (Fraser et al., 2004), and biotite (Vavra et al., 1996). Zircon coronae have been reported around Martian baddeleyite as a result of shock metamorphism (Moser et al., 2013). In mafic metamorphic rocks, precipitation of zircon is commonly associated with the Fe–Ti oxides (Bingen et al., 2001; Ewing et al., 2013, 2014) due to similar chemical properties of Zr and Ti.

Zirconium and titanium both belong to group 4 in the periodic table, have close chemical properties and are usually regarded as relatively immobile trace elements (e.g., Mohamed and Hassanen, 1996). In the group of incompatible cations, Zr and Ti belong to high field strength (HFS) elements, which are smaller and are highly charged compared with large ion lithophile (LIL) elements. The chemical similarities result in a positive correlation between Zr and Ti for most rock suites and in their ability to replace each other in oxides (e.g., Morisset et al., 2005). The fact that Zr oxides and Ti oxides are spatially related in many rocks confirms chemical similarities between Zr and Ti.

Thus, rutile and ilmenite are the main minerals interpreted to influence the Zr mass balance in metabasites in the absence of other Zr phases (e.g., Ferry and Watson, 2007; Tomkins et al., 2007; Morisset and Scoates, 2008; Ewing et al., 2013). In the absence of zircon, rutile can be the main phase holding Zr and Hf in the absence of zircon (Ewing et al., 2014). Zirconium is a common component of rutile, in which its content can reach 10 000 ppm (e.g., Ewing et al., 2013); thus, Zr distributions generally reflect the formation and decomposition of rutile. The temperature dependence of Zr solubility in rutile can have a fundamental impact on the zircon growth rate (Kohn et al., 2015) and controls zircon stability (Kelsey and Powell, 2011). The zirconium-in-rutile thermometer for the

rutile–quartz–zircon system was calibrated by a number of authors (e.g., Watson et al., 2006; Ferry and Watson, 2007; Tomkins et al., 2007; Lucassen et al., 2010; Ewing et al., 2013), who have shown a large temperature- and pressure-dependent solubility of Zr in rutile. Zircon growth is frequently associated with the oxide transition from Zr-rich rutile to ilmenite during late-stage exhumation and cooling under a large variety of $P - T$ conditions (Ewing et al., 2013). Magmatic and metamorphic ilmenite can also contain significant amounts of Zr (Bingen et al., 2001; Morisset et al., 2005; Charlier et al., 2007), up to more than 500 ppm (e.g., Morisset and Scoates, 2008). Consistently, many authors describe zircon precipitation on ilmenite (e.g., Bingen et al., 2001; Austrheim et al., 2008; Morisset and Scoates, 2008) (see below).

1.3 Occurrences of fine-grain zircon and zircon corona textures

In igneous rocks zircon usually forms euhedral elongated single crystals that are shaped by a combination of prismatic and pyramidal faces, whereas metamorphic zircon is characterized by roundish or irregular shapes (Corfu et al., 2003). Rarely, zircon has unusual saccharoidal or needle-shaped morphology or forms coronae (Corfu et al., 2003 and references therein). Mineral–fluid interactions, decomposition of Zr-bearing minerals, and exsolution from Zr-bearing accessory and rock-forming minerals can result in such unusual zircon textures (e.g., Corfu et al., 2003 and references therein; Dempster et al., 2004, 2008; Rasmussen, 2005), even at low metamorphic grades (e.g., Dempster et al., 2008).

In natural samples, there are several documented examples of zircon coronae textures from igneous and metagneous rocks (e.g., Bingen et al., 2001; Söderlund et al., 2004; Austrheim et al., 2008), as well as from metapelites of the IVZ (Pape et al., 2016). Such textures are found in rocks of different metamorphic grades, ranging from prehnite–pumpellyite to eclogite facies (Austrheim et al., 2008). It has been suggested that coronae textures may evolve in magmatic rocks as a result of slow cooling (Morisset et al., 2005) and in metamorphic rocks due to mineral–fluid reactions or exsolution with fluid-aided diffusion along grain boundaries during progressive metamorphism (e.g., Bingen et al., 2001).

One of the first descriptions of zircon coronae in mafic metagneous rocks was done by Söderlund et al. (2004). The authors attributed formation of secondary fine-grained zircon to the breakdown of baddeleyite in the presence of silica (saccharoidal zircon) and to consumption of minerals that have trace amounts of Zr, such as ilmenite (coronitic zircon). Both of these textural types of secondary zircon precipitated under prograde heating. Bingen et al. (2001) reported hat-shaped zircon grains and coronae around ilmenite in granulites and amphibolites. Charlier et al. (2007) and Austrheim et al. (2008) reported fine-grained zircon chains around, but at a distance from ilmenite and rutile grains in metagabbros.

These authors suggested that zircon chains had grown around primary Fe–Ti oxides and, therefore, trace the former grain boundaries. Fine-grained zircon was reported to frame some rutile grains in the metapelitic septae from the IVZ (Ewing et al., 2013; Pape et al., 2016). Morisset and Scoates (2008) reported 1–100 μm thick zircon coronae around ilmenite in mafic plutonic rocks. They consider it to be a result of Zr diffusion from ilmenite during slow cooling, aided by hydrothermal fluid.

In this study, we report the two textural types of zircon coronae, characterized by various thickness and aspect ratio, occurring within Fe–Ti oxides in granulitic metapelites. We present evidence that these textures formed as a product of the breakdown of Fe–Ti oxides, which helps to understand the initial mineral paragenesis of the host rock and reveals former reaction fronts.

2 Geological background and sampled locality

The IVZ in the Southern Alps (northern Italy) consists of a NE–SW trending, steeply dipping sequence of metasedimentary and metagneous basic rocks, ultrabasic mantle tectonites and a large underplated mafic igneous complex (Fig. 1a) (e.g., Brodie and Rutter, 1987; Brodie et al., 1992; Rutter et al., 2007). The sequence predominantly consists of metasedimentary rocks in the SE and metabasic rocks and strongly depleted metapelites in the NW. Metamorphic grade increases progressively from amphibolite facies in the SE to granulite facies in the NW. The IVZ is generally accepted as a section through the lower continental crust that experienced regional metamorphism during the uppermost Paleozoic and was tectonically overturned and uplifted. The IVZ is delimited by the Insubric Line in the NW and the Pogallo Line in the SE (Brodie and Rutter, 1987; Barboza et al., 1999; Rutter et al., 2007; Quick et al., 2009).

The sampled outcrop near the village Cuzzago (Val d'Ossola) shows massive, non-foliated granulite-facies metasediments, known as strolalites. Strolalite is defined as granulite-facies metapelite, consisting of garnet, sillimanite and biotite with leucocratic patches and veins, composed of quartz, plagioclase and K-feldspar (Bea and Montero, 1999) or as granoblastic graphite–sillimanite–garnet gneiss, one of the components of the IVZ septa (Barboza et al., 1999). Local foliation and/or compositional layering of strolalites is moderately folded (e.g., Kovaleva et al., 2014, their Fig. 1C). Strolalites are broken by orthogonal sets of fractures and crosscut by a contrasting layer of darker gneiss (45°59'46.46" N, 8°21'38.65" E, sampled rock; Fig. 1b), which is macroscopically massive to weakly foliated, broken by abundant faults normal to foliation. The foliation of the layer strikes NW (310°, angle 77°) and the lineation plunges to the NE (34° towards 038°). No obvious kinematic indicators were observed in the host strolalites or in the sampled dark gneiss. However, detailed structural in-

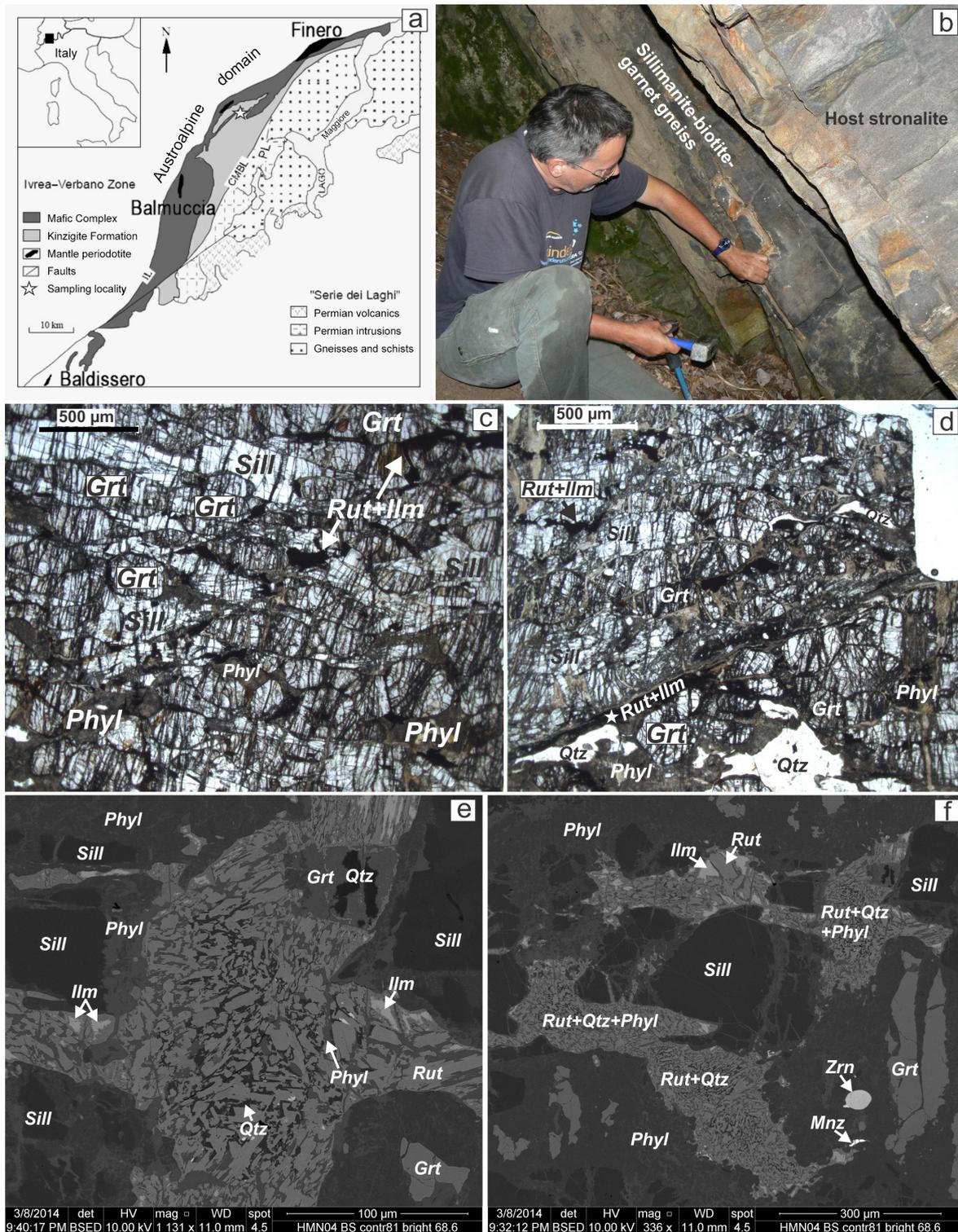


Figure 1. (a) Geological map of the Ivrea-Verbano Zone after Zanetti et al. (1999) with the sampling location indicated by a star. (b) Field photograph of the sampled outcrop with the dyke-shaped body of the sillimanite-biotite-garnet gneiss, interpreted as restitic, hosted by mylonitized and fractured stromalite. (c-d) Plain-polarized light photomicrographs. Two generations of veins are visible: black veins and pockets (Fe-Ti oxides) and brown-grey material (mixture of fine-grained phyllosilicates and K-feldspar). White star in panel (d) indicates the position of Fig. 4a. (e-f) BSE images with mineral paragenesis. Note in panel (e) that the rutile aggregate contains ilmenite cores (bright grey) and forms intergrowths with two different phases: phyllosilicates from the reaction rim (grey shade, slightly darker than rutile) and quartz (the darkest phase), all indicated by arrows. Sill: sillimanite, Grt: garnet, Qtz: quartz, Ilm: ilmenite, Rut: rutile, Rut + Qtz: rutile-quartz intergrowths, Phyl: fine-grained mixture of phyllosilicates, Zrn: zircon detrital grain, and Mnz: monazite.

vestigations of the shear zones in the neighboring Val Strona revealed numerous structures that provide consistent evidence of sinistral shear (Siegesmund et al., 2008, and references therein).

The sampled gneiss consists of sillimanite–biotite–garnet intergrowths (Fig. 1c–d). Such restitic mineral assemblage in granulitic metapelites is interpreted to form due to partial melting and separation of leucosome (e.g., Barboza et al., 1999; Luvizotto and Zack, 2009; Ewing et al., 2013; Pape et al., 2016). Thus, the sampled gneiss is a restite, resulting from migmatization at peak granulite-facies conditions (e.g., Ewing et al., 2013). Partial melting is also responsible for apparent layering of host strolonites (Bea and Montero, 1999; Siegesmund et al., 2008), which are composed of alternating leucocratic and melanocratic layers.

The investigated sample came from the northwestern part of the IVZ, where metapelites and metagabbro were re-equilibrated under granulite-facies conditions prevailing during crustal attenuation–extension and contemporaneous magmatic underplating between 315 and 270 Ma (Rutter et al., 2007; Quick et al., 2009; Sinigoi et al., 2011; Klötzli et al., 2014). P – T estimates in the neighboring valleys, Val Strona and Val d’Ossola, indicated granulite-facies P – T metamorphic conditions in the metapelites at a maximum of 750 ± 50 °C and 0.6 ± 0.1 GPa (Sills, 1984). Zr-in-rutile temperatures of up to 850–930 °C were obtained for granulite-facies metapelites from Val d’Ossola (Luvizotto and Zack, 2009).

Ubiquitous faulting of restitic material and both faulting and folding of host strolonite is due to intensive deformation taking place in granulitic metapelites during a long-time span after peak metamorphism (Siegesmund et al., 2008). According to Siegesmund et al. (2008) brittle and ductile deformation acted simultaneously during formation of shear zones, and their close interactions resulted in complex deformation microstructures.

3 Sample preparation and analytical methods

Zircon textures have been examined in situ using polished thin sections that were mechanically prepared with a final polish using 0.25 µm diamond paste. Zircon grains were identified by backscattered-electron (BSE) imaging and were additionally characterized by cathodoluminescence (CL) imaging for the internal growth features using an FEI Inspect S scanning electron microscope equipped with a Gatan MonoCL system (Faculty of Earth Sciences, Geography and Astronomy, University of Vienna, Austria). Imaging conditions were at 10 kV accelerating voltage, CL image resolution of 1500×1500 to 2500×2500 pixels using a dwell time of 80.0–150.0 ms and a probe current of 4.5–5.0 nA. Qualitative chemical compositions of host phases to zircon were made using an energy-dispersive X-ray (EDX) spectrometer. Orientation contrast images of zircon grains (e.g.,

Fig. 2) were taken using a forescatter electron (FSE) detector on a chemically polished sample surface. The FSE detector is mounted on the electron backscatter detector (EBSD) tube of a FEI Quanta 3D field emission gun (FEG) instrument (Faculty of Earth Sciences, Geography and Astronomy, University of Vienna, Austria), which is equipped with a Schottky field emission electron source. Electron beam conditions were 15 kV accelerating voltage and 2.5–4 nA probe current using the analytic mode. Stage settings were at 70° tilt and 14–16 mm working distance. Full quantitative chemical compositions of host minerals (Tables 1–2) were determined by Cameca SX 100 electron microprobe equipped with four wavelength-dispersive spectrometers (WDSs) and an EDX system for high quality of quantitative chemical analyses (Faculty of Earth Sciences, Geography and Astronomy, University of Vienna, Austria). Operating conditions were 15 kV accelerating voltage and 100 nA probe current. The detection limits in parts per million (ppm) for each microprobe analysis point are presented in Table S1 in the Supplement.

4 Results

4.1 Microscopic description

The generally restitic mineralogy of the sample is composed of garnet, biotite and sillimanite with minor amounts of cordierite, ilmenite, rutile, K-feldspar and quartz (Fig. 1c–f). The primary mineralogy indicates prograde–peak mineral paragenesis, which consists of biotite, sillimanite and garnet. The foliation is formed by a fabric of elongated garnet and sillimanite crystals 0.5–1 mm in length that compose 80–90 % of the sample (Fig. 1c–d). The stretching lineation is formed by elongated biotite crystals. Biotite contains numerous micrometer-sized apatite needles and is mostly replaced by chlorite. Primary metamorphic fabric is crosscut by several generations of veins and/or fractures (Fig. 2c–f), which were formed during cataclastic deformation and shear zone development (e.g., Siegesmund et al., 2008). Fractures are filled with post-peak and late hydration mineral assemblages. The earlier generation of veins is mostly composed of Fe–Ti oxides and their intergrowths with quartz (Fig. 1c–d, black material). Fe–Ti oxides form aggregates with lobate boundaries with the primary minerals (garnet and sillimanite) (Fig. 1e–f). The network of veins of the later generation cross-cuts the veins of the earlier generation or follows their contacts. These later veins are more abundant than earlier ones and are composed of fine-grained phyllosilicates, such as chlorite, muscovite and/or phengite, and may also contain K-feldspar patches in the vein cores (Fig. 1c–d, grey-brown material; Table 1). Abundance of phyllosilicates indicates post-metamorphic hydration reactions. Large (up to 2 mm in length and 0.3 mm thick) elongate quartz aggregates generally follow the vein distribution (Fig. 1d, bottom part).

Table 1. Results of the microprobe analyses of the rock-forming silicates; n.d.: not detected. Grt: garnet, Bt: biotite, Chl: chlorite, Phyl: phyllosilicate(s), Pheng: phengite, and Mus: muscovite.

Mineral	SiO ₂	TiO ₂	Al ₂ O ₃	MgO	CaO	MnO	FeO	BaO	Na ₂ O	K ₂ O
Grt core	38.90	0.02	22.22	9.33	1.51	0.31	28.60	n.d.	n.d.	n.d.
Grt core	38.14	0.01	21.69	6.19	1.46	0.43	32.92	n.d.	n.d.	n.d.
Grt core	38.16	0.01	21.70	6.37	1.46	0.49	32.51	n.d.	0.01	n.d.
Grt core	39.01	n.d.	22.15	9.19	1.51	0.36	28.52	n.d.	0.02	n.d.
Grt core	39.10	0.02	22.09	9.26	1.50	0.32	28.80	n.d.	0.02	n.d.
Grt core	38.92	0.03	22.11	9.13	1.46	0.30	28.84	n.d.	0.03	n.d.
Grt core	38.83	0.01	22.00	8.34	1.41	0.35	29.92	n.d.	0.03	n.d.
Grt core	37.80	0.02	22.03	6.75	1.49	0.55	31.69	n.d.	0.01	n.d.
Grt rim	38.81	0.01	22.12	8.58	1.54	0.34	29.44	n.d.	0.01	n.d.
Grt rim	38.81	0.03	22.06	8.28	1.53	0.34	30.31	n.d.	0.03	n.d.
Grt rim	38.95	0.01	21.82	7.73	1.52	0.40	30.77	n.d.	n.d.	n.d.
Grt rim	38.35	0.01	21.95	7.77	1.51	0.42	30.35	n.d.	0.02	n.d.
Grt rim	38.61	0.01	21.59	5.67	1.46	0.52	33.25	n.d.	0.01	n.d.
Grt rim	38.39	0.07	21.78	5.96	1.46	0.53	32.82	n.d.	n.d.	n.d.
Grt rim	37.93	0.21	21.49	5.65	1.52	0.55	32.93	n.d.	0.02	n.d.
Grt rim	38.30	0.07	21.86	6.82	1.45	0.43	31.69	n.d.	n.d.	n.d.
Grt rim	38.34	0.04	21.75	7.15	1.48	0.43	31.37	n.d.	0.01	n.d.
Grt rim	37.35	0.01	21.12	3.85	1.50	0.87	35.51	n.d.	n.d.	n.d.
Grt rim	37.38	0.03	21.12	3.86	1.45	0.81	35.22	n.d.	n.d.	n.d.
Grt rim	37.65	n.d.	21.30	3.98	1.43	0.78	35.34	n.d.	n.d.	n.d.
Bt core	37.15	6.27	15.51	15.79	0.03	0.03	10.75	n.d.	0.11	9.83
Bt core	37.15	1.32	16.53	18.79	0.13	0.03	11.36	n.d.	0.10	9.16
Bt rim	34.81	1.74	17.17	14.41	0.04	n.d.	17.50	n.d.	0.12	7.83
Bt rim	35.22	2.82	16.59	13.55	0.02	0.04	18.27	n.d.	0.10	8.15
Bt rim	35.61	1.46	16.14	14.38	0.16	0.02	17.94	n.d.	0.09	7.38
Bt rim	34.75	1.27	17.16	13.37	0.03	0.04	19.60	n.d.	0.08	7.27
Chl over Bt	29.75	0.03	18.37	16.40	0.10	0.05	22.91	n.d.	n.d.	0.14
Chl over Bt	27.72	0.01	19.47	15.12	0.08	0.06	25.34	n.d.	n.d.	0.05
Chl over Bt	27.29	0.21	19.76	15.84	0.03	0.07	24.21	n.d.	n.d.	0.03
Chl over Bt	27.64	0.28	20.07	16.34	0.04	0.04	23.41	0.04	n.d.	0.04
Chl over Bt	28.09	0.40	19.07	15.97	0.06	0.06	24.07	0.04	n.d.	0.20
Chl new	27.14	0.02	21.48	14.10	0.03	0.09	24.90	n.d.	n.d.	0.07
Chl new	24.84	0.03	22.94	12.98	0.05	0.06	26.44	n.d.	n.d.	0.06
Chl new	28.47	0.02	18.16	15.53	0.06	0.05	24.92	n.d.	0.01	0.06
Chl new	29.00	0.08	18.92	15.73	0.09	0.06	24.07	n.d.	0.02	0.21
Chl new	29.44	0.06	19.27	15.65	0.07	0.06	24.44	n.d.	0.02	0.26
Chl new	28.70	0.02	18.38	15.42	0.08	0.07	25.45	0.05	0.04	0.09
Chl new	29.30	0.04	17.69	15.27	0.08	0.06	25.39	0.02	0.01	0.10
Chl new	31.95	4.44	17.50	11.51	0.04	0.05	21.81	0.04	0.04	4.64
Phyl matrix	44.34	n.d.	31.75	3.84	0.13	n.d.	6.06	0.28	0.33	7.70
Phyl matrix	49.04	0.01	32.24	1.68	0.04	n.d.	1.63	0.17	0.18	10.58
Phyl matrix	45.92	0.01	32.36	2.78	0.12	n.d.	4.12	0.47	0.32	8.63
Phyl matrix	43.30	0.18	32.40	4.12	0.15	0.05	7.16	n.d.	0.33	7.12
Phyl matrix	47.74	0.12	33.88	1.98	0.17	0.01	3.05	n.d.	0.42	8.67
Phyl matrix	44.89	0.21	32.84	3.28	0.10	n.d.	4.87	n.d.	0.37	8.39
Phyl matrix	44.91	0.02	32.38	3.38	0.17	0.02	5.17	n.d.	0.38	7.81
Phyl matrix	38.78	0.05	28.79	6.94	0.10	0.02	12.02	0.19	0.27	4.91
Phyl matrix	45.22	0.03	32.53	3.29	0.09	0.03	4.49	0.73	0.37	8.28
Pheng/Mus	47.45	0.07	35.68	1.31	0.09	n.d.	1.65	0.32	0.40	9.53
Pheng/Mus	48.41	0.17	35.13	1.34	0.14	n.d.	1.46	0.37	0.43	9.19
Pheng/Mus	46.62	0.12	35.34	1.43	0.10	0.01	1.94	0.36	0.45	9.07

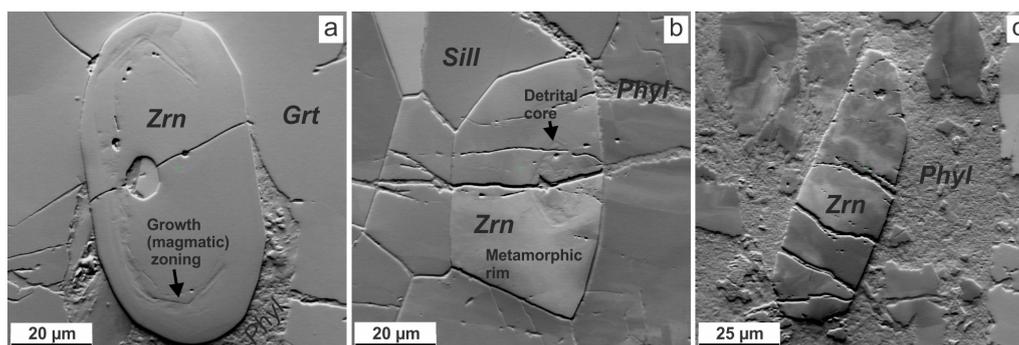


Figure 2. Orientation contrast images of detrital zircon in the sampled gneiss: (a) zircon grain hosted by garnet; note the concentric growth zoning. (b) Zircon grain hosted by sillimanite, note the small detrital core (right hand side) and wide metamorphic rim. (c) Zircon grain hosted by a fine-grained matrix that fills the veins; note intensive change in orientation contrast, especially conspicuous in the upper part of the grain. Orientation contrast image indicates the crystal-plastic deformation of the zircon grain and surrounding mineral fragments (garnet and sillimanite). Fracture surfaces appear to be dissolved. Mineral abbreviations as in Fig. 1.

Table 2. Results of the microprobe analyses of the Fe–Ti oxides, n.d.: not detected. Ilm: ilmenite and Rut: rutile.

Mineral	Ta ₂ O ₅	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	Nb ₂ O ₃	MgO	MnO	FeO	NiO	Total
Ilm	0.04	0.03	52.82	0.02	n.d.	0.11	0.05	0.85	45.18	n.d.	99.15
Ilm	0.03	0.14	53.07	0.12	0.02	0.12	0.07	0.85	44.30	0.02	98.73
Ilm	n.d.	0.15	53.72	0.12	n.d.	0.04	0.03	0.89	43.56	n.d.	98.52
Rut	n.d.	0.60	98.31	0.22	0.08	0.25	0.04	n.d.	0.44	n.d.	99.92
Rut	n.d.	0.80	98.46	0.26	0.08	0.40	0.01	n.d.	0.40	n.d.	100.42
Rut	n.d.	2.10	93.19	1.42	0.06	0.22	0.60	n.d.	1.46	n.d.	99.03

Veins and fractures form a conjugated orthogonal network, stretching in at least two directions in a 2-D section.

Accessory minerals are zircon and monazite (e.g., Figs. 1f, 2). Where hosted by garnet, zircon forms roundish elongated crystals with aspect ratios from 1 : 1 to 1 : 3 and lengths from 30 to 100 µm (Fig. 2a). Where forming intergrowths with sillimanite, zircon reveals well-developed faces and forms triple junctions with the adjacent sillimanite grains (Fig. 2b), reflecting equilibration growth with sillimanite during prograde and peak metamorphism. Where hosted by fine-grained material that fills fractures, zircon crystals are elongated, with an aspect ratio from 1 : 2 to 1 : 3; these grains are fractured and fragmented. The fragments have irregular dissolved boundaries and show evidence of crystal-plastic deformation (Fig. 2c). Vermicular- and hat-shaped zircon aggregates and zircon coronae are spatially associated with each other and occur within ilmenite–rutile–quartz or rutile–quartz clusters and/or intergrowths (Figs. 3–5), which fill transgranular fractures and pockets in the gneiss (e.g., Fig. 4a).

4.2 Zircon microstructures and textures

Zircon textures reported in this study are coronae, by which we mean thin envelopes or shells in 3-D. Accordingly, in the 2-D plane of a sample they have thread- or worm-like shapes

(depending on the thickness and aspect ratio). Zircon coronae in our sample occur as two main textural types. The first type is referred to as vermicular-shaped (coarser-grained) aggregates, which have a thickness ≥ 5 µm and an aspect ratio of 1 : 4 to 1 : 20. The second type is referred to as coronae (finer-grained) zircon aggregates, which have a thickness ≤ 1 µm and an aspect ratio of approximately 1 : 100 (e.g., Figs. 3–4). There is also a third (subordinate) coronae type: hat-shaped aggregates that are the result of zircon coronae overgrowth preexisting (probably detrital) grains.

4.2.1 Vermicular textures

This textural type occurs as lamellae-like intergrowths with rutile. Some vermicular-shaped zircon grains are hosted by thin rutile-quartz intergrowths, in which rutile forms < 1 µm thin and 1–3 µm long needles (Fig. 3b, matrix). Such needle shapes are evidence of rapid rutile recrystallization and re-equilibration during the metamorphic evolution.

Figure 3a shows a zircon aggregate composed of three large vermicular-shaped grains (indicated by V, enlarged in Fig. 3b, c, e). These vermicular grains are 5 to 15 µm thick and 20 to 50 µm long and have diffuse or auroral-light (Corfu et al., 2003) CL zoning (Fig. 3b, c, e). Vermicular-shaped grains have curved (Fig. 3b) or ragged (Fig. 3e) boundaries, a crescent-like shape (Fig. 3c, e) and are commonly broken

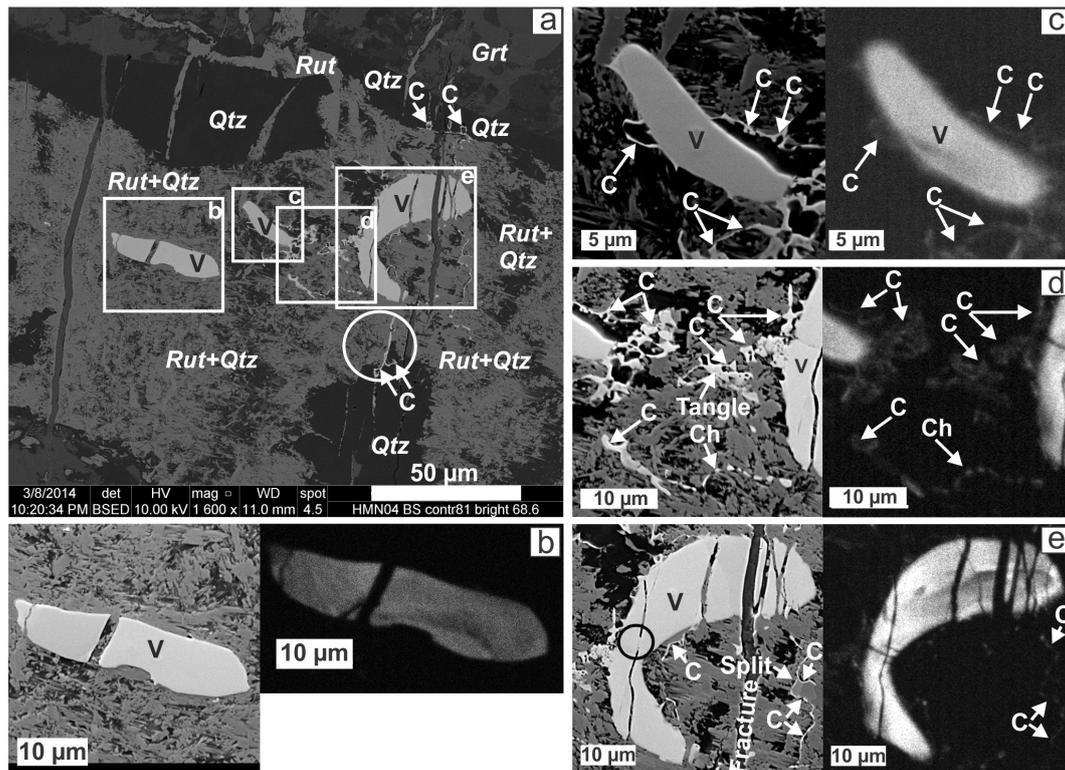


Figure 3. (a) BSE image of the vermicular-shaped zircon aggregates (“V”) and zircon coronae (“C”). Arrows indicate zircon coronae that trace the quartz–rutile boundary or fill the cavities in quartz; the circle highlights a corona that fills the fracture. Mineral abbreviations as in Fig. 1. (c–e) Enlarged BSE (left) and CL (right) images of the areas indicated in panel (a). V highlights the vermicular-shaped zircon grains and C points to zircon coronae (the difference between V and C is in thickness). “Ch” in panel (d) points to the chain of submicron-sized zircon grains, and “Tangle” points to the tangled occurrence of coronae. Arrows in panel (e) indicate the directions of the reaction fronts, and “Split” points to the branching of zircon coronae. The circle in panel (e) highlights a partially healed fracture in vermicular zircon.

with transgranular fractures (Fig. 3b, e). Some of these fractures are traced in the host rock and filled with fine-grained phyllosilicates (Fig. 3e), which suggests that vermicular zircon predates the cataclasis and hydration with phyllosilicate growth. Furthermore, some fractures in vermicular grains (Fig. 3e) are partially healed by low-CL zircon material. This indicates that some precipitation of zircon has occurred after fracturing.

Another example of vermicular-shaped zircon aggregate is presented in Fig. 4. This texture is found in a vein of the early generation filled by the rutile–quartz intergrowths and elongate aggregates of quartz (Figs. 1d, 4a). A large zircon aggregate has a W shape and consists of two major fragments (Fig. 4b). The thickness of the vermicular zircon varies from 5 to 20 μm , and the total length is about 200 μm . The W-shaped vermicular aggregate shows diffuse CL zoning (Fig. 4d). The lower part of this aggregate used to extend to the right (Fig. 4c) and connect with the smaller vermicular grain at the right-hand side from the W-shaped grain (Fig. 4a). This 50 μm long extension was removed by subsequent polishing. The lower right tip of the aggregate drops below the surface plane of the thin section. The CL image

and the EDX map of Zr distribution reveal the blurred trace around the lower right tip (Fig. 4d–e, grey arrows). This indicates that the zircon aggregate continues deeper into the sample at a shallow angle, and its signal is documented by CL and EDX from a few micrometers below the surface. As such, the aggregate represents an envelope in 3-D. The W-shaped zircon grain is plastically deformed in its central part, which is indicated by an orientation contrast image (Fig. 4f). Rotation of the lattice reaches 7° with respect to the undeformed lattice (Kovaleva et al., 2016), indicating that this vermicular zircon grain predated shearing and ductile deformation.

Vermicular aggregates presented in Figs. 3 and 4 are associated with coronae textures, unlike aggregates in Fig. 5a. These aggregates are fractured and hosted by a rutile–phyllosilicate aggregate, which fills the pocket between sillimanite and garnet (V in Fig. 5a).

4.2.2 Coronae textures

The matrix around some vermicular grains (Figs. 3c–e, 4c) contains abundant and continuous thin zircon coronae (C in Figs. 3a, c–e, 4a–e, 5) and fine-grained zircon chains (Ch in

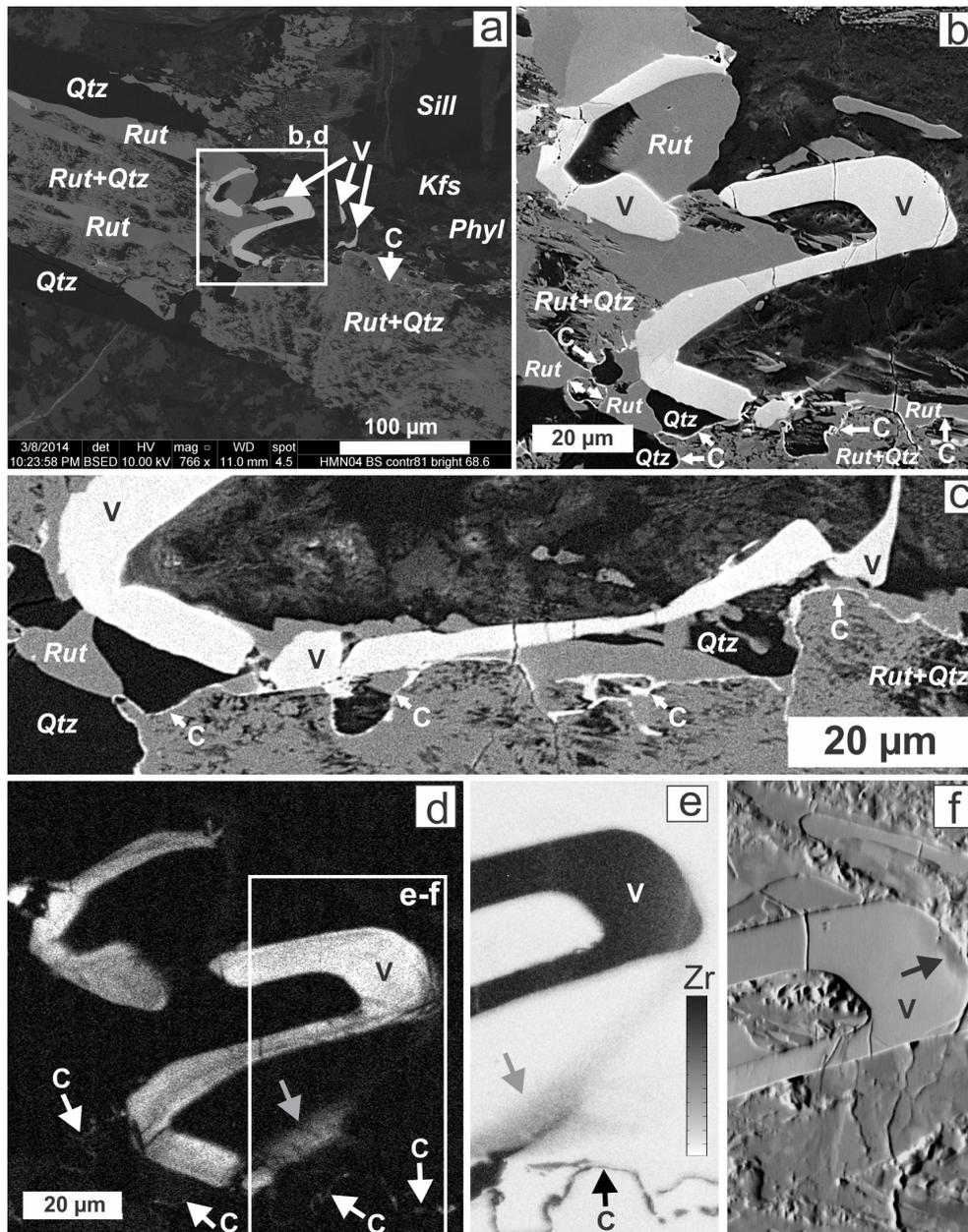


Figure 4. (a) BSE image of the zircon aggregate, which forms intergrowth with rutile and quartz. White rectangle highlights the area enlarged in panels (b) and (d). (b) Enlarged BSE image of the area indicated in panel (a). Arrows point out the direction of reaction fronts. (c) Enlarged area of the lower part of panel (b). The middle part of vermicular texture (“V”) is present (it was subsequently polished away and thus absent in panels (a–b)). Arrows point out the direction of the reaction front. (d) Enlarged CL image of the area indicated in panel (a). White arrows as in panel (b); grey arrow points to the wedged zircon texture that continues below the surface. (e) Qualitative EDX intensity map for Zr of the area indicated in panel (d). Black arrows point out the direction of reaction front; grey arrow as in panel (d). (f) Orientation contrast image of the area indicated in panel (d). Arrow points to plastically deformed tip of vermicular zircon grain. V in panels (a–f) highlights vermicular zircon grains and C highlights zircon coronae. Mineral abbreviations as in Fig. 1; Kfs: K-feldspar.

Figs. 3d, 4e). Coronae are $\leq 1 \mu\text{m}$ thick and are up to $200 \mu\text{m}$ long. They can be observed anywhere around and within rutile–quartz–phyllosilicates, ilmenite–rutile and ilmenite–rutile–quartz aggregates (Figs. 3–5). Coronae are distributed rather randomly and are commonly found in the presence of

larger zircon grains, which can be vermicular-shaped aggregates (e.g., Figs. 3–4), detrital grains (e.g., Fig. 5a, b, d) or peak metamorphic grains (e.g., Fig. 5c). At the same time, not all rutile–quartz intergrowths (Fig. 1e–f, 5a) and not all

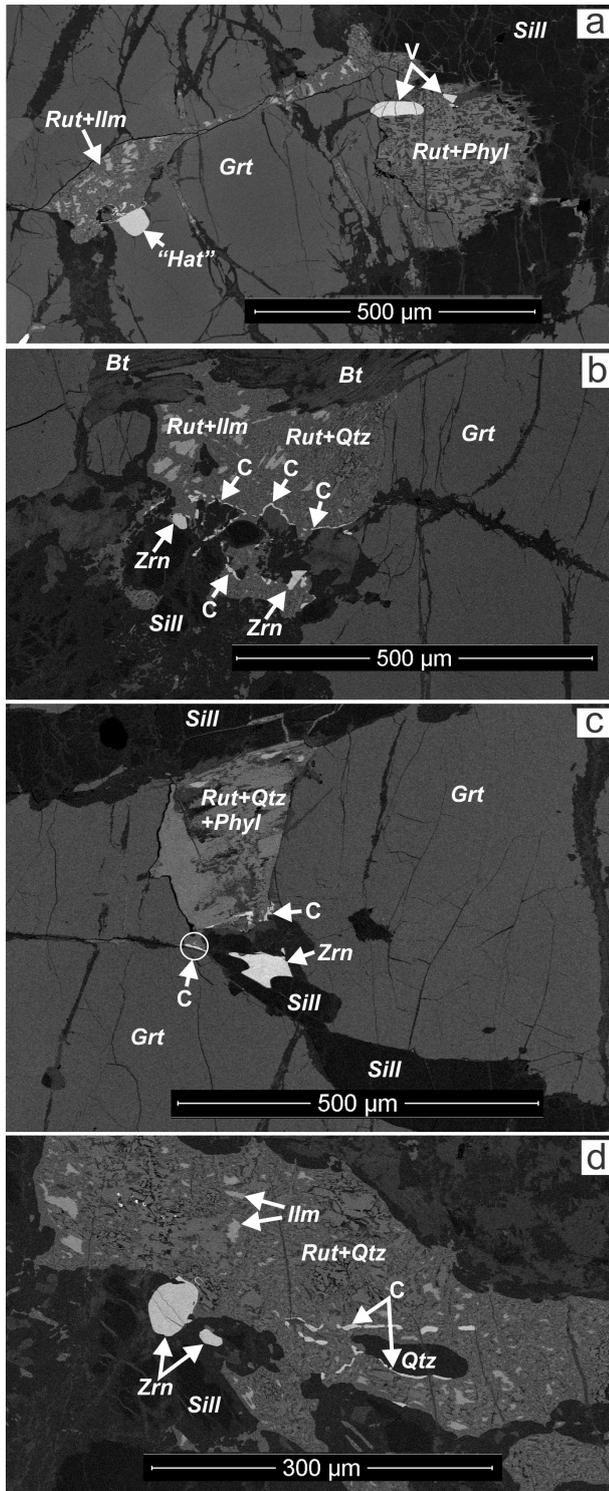


Figure 5. BSE images of mineral reactions that contain detrital zircon grains and associated zircon coronae. Mineral abbreviations are as in Fig. 1: Bt: biotite; Hat: hat-shaped zircon aggregate. “V” highlights the vermicular zircon grains, “C” points to the zircon coronae, and Zrn is the preexisting detrital and metamorphic zircon grains. Arrows in panel (b) show the direction of the mineral reaction front.

vermicular-shaped grains (V in Fig. 5a) are associated with zircon coronae.

Both zircon coronae and fine-grained chains have distinguishable CL responses (Figs. 3c–e, 4d). Coronae occur as continuous threads that form splits (Figs. 3e, 4e) or isolated tangles (Fig. 3d). An especially dense network of zircon coronae is presented in Fig. 3d. Coronae are commonly attached to the larger vermicular zircon grains (Figs. 3c–e, 4b–c). Some coronae follow the phase boundaries between rutile–quartz aggregates and quartz (Figs. 3a, 4c, 5d) or rutile–quartz aggregates and sillimanite (C in Fig. 5b–c). Some thin zircon coronae extend outside of rutile–quartz intergrowths and fill fractures in quartz and garnet (Figs. 3a and 5c accordingly, circles), which suggests zircon coeval-to post-fracturing precipitation.

4.2.3 “Hat” textures

The rutile–ilmenite intergrowths, adjacent to a rectangular zircon grain, fill a fracture in garnet adjacent to a rectangular zircon grain (Fig. 5a). Zircon coronae trace the boundary between garnet and rutile–ilmenite intergrowths and are connected to the rectangular zircon grain, so that the latter acquires a hat-like shape (Hat in Fig. 5a, after Bingen et al., 2001). Another example of similarly formed zircon aggregate does not have such a well-developed hat shape. It is a roundish zircon grain (Zrn in Fig. 5d), which is spatially associated with coronae and has short coronae outgrowths pointed towards the rutile–quartz aggregate.

4.3 Microprobe data

Mineral electron microprobe data are presented in Tables 1 and 2. These data are used to determine the temperature of metamorphism using the garnet–biotite thermometer and to support the suggestion of possible mineral–fluid reactions. The X_{Fe} of garnet is systematically lower in the cores than in the rims and in the smaller fragments. The same applies to the biotite (Table 1). Garnet rims are also systematically enriched in Mn, compared to the cores. Compositions of Fe–Ti oxides (Table 2) demonstrate that rutile is much higher in SiO_2 content than ilmenite. Rutile is also slightly enriched in such elements as Al, Cr and Nb, but lower in Mn (Table 2) compared to ilmenite.

5 Discussion

5.1 Mineral reactions

Mineral textures in the studied sample provide important information about the reactions that could have enhanced the growth of zircon coronae during metamorphism. We suggest the following reaction sequence: (1) formation of peak metamorphic phases and partial melting of the metapelites; (2) initial resorption of peak metamorphic phases and crys-

tallization of interstitial ilmenite with lobate boundaries in fractures; (3) retrograde metamorphism, further resorption and fracturing of the high-temperature phases, hydration reactions with formation of phyllosilicates and decomposition of ilmenite to rutile; and (4) further cooling and recrystallization of rutile.

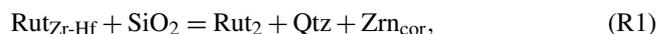
Temperature estimations were done using the garnet–biotite thermometer using microprobe data (Table 1). Various calibrations of this geothermometer (Thompson, 1976; Holdaway and Lee, 1977; Ferry and Spear, 1978; Hodges and Spear, 1982; Perchuk and Lavrent'eva, 1983; Bhattacharya et al., 1992) gave temperatures for garnet–biotite preserved cores of 570–700 °C, for inner rims 800–860 °C, and for outer rims 820–1090 °C. Estimations were done for pressures of 0.7 and 1.0 GPa. Pressure variations did not have any significant effect on the resulting temperatures. It is, however, possible, that garnet and biotite rims were affected by diffusion from the host environment during retrograde metamorphism. Mineral textures (Figs. 1e–f, 2c, 4a, 5) and microprobe analyses (Table 1) indicate that the initial granulite-facies garnet, biotite and sillimanite were intensely altered and resorbed. The fragments of garnet and sillimanite have ragged edges and are plastically deformed, dissolved and altered. Mn and Fe, enriched in the rims of large garnet grains and in small garnet fragments, suggest garnet resorption and hydration during retrograde metamorphism (e.g., Tuccillo et al., 1990) after peak metamorphism. Thus, garnet rims do not indicate peak metamorphic temperatures; therefore, the rim temperatures are erroneous. More likely, peak metamorphic temperatures in this IVZ section were between 700 and 860 °C (temperatures obtained for the cores and inner rims), in agreement with previous estimations (Sills, 1984; lowermost estimations of Luvizotto and Zack, 2009).

Pockets and fractures in garnet and sillimanite are filled with Fe–Ti oxide aggregates with lobate boundaries. Thus, ilmenite probably crystallized from the partial melt and/or fluid after the formation of peak metamorphic phases and after their resorption was initiated. Occasionally, veins with Fe–Ti oxides are associated with the quartz aggregates; thus, quartz possibly formed in the early generation of veins together with ilmenite. Ubiquitous fracturing of the rock (e.g., Fig. 1c–d) and crystal-plastic deformation of zircon (Figs. 2c, 4f) indicate the extreme conditions of post-peak metamorphism deformation (e.g., Siegesmund et al., 2008).

Further retrograde (greenschist facies) metamorphism led to hydration reactions and formation of veins filled with phyllosilicates and K-feldspar. The following features are regarded as evidence of intensive mineral–fluid reactions in the dry restitic granulite-facies rock (e.g., Rajesh et al., 2013): reaction rims around fragments of granulite-facies minerals (e.g., Fig. 1e), fine-grained phyllosilicate mixture that fills fractures (e.g., Fig. 1c–d, f), quartz veins (Figs. 1d, 4a), alteration of biotite with chloritization and exsolution of apatite needles. Water-rich fluids could have been sourced from the decomposing biotite (e.g., Pape et al., 2016). Rare il-

menite cores are surrounded by rutile rims (Figs. 1e–f, 5a–b, d). Thus, post-peak, trace-element-rich ilmenite was partially or entirely decomposed to rutile, which resulted in the migration of excess Fe into the matrix and into the garnet and biotite rims. Fe from ilmenite and Mg diffusing out of garnet and biotite rims are needed to compensate for the formation of the large volume of Mg–Fe phyllosilicates in the second generation of veins (Figs. 1c–d, 2c; Table 1). K from biotite and Al from sillimanite would allow and/or favor the growth of K-feldspar in the veins (e.g., Fig. 4a). Excess Hf and Zr from breaking down ilmenite are responsible for the formation of zircon intergrowths with rutile (thick vermicular-shaped grains, e.g., Figs. 3b, c, e, 4a–c). Newly formed rutile is enriched in trace elements, possibly due to the decreased volume of Fe–Ti oxides (e.g., Austrheim et al., 2008). This rutile is also enriched in silica (Table 2) that requires sourcing SiO₂ from the environment and may indicate solid solution of SiO₂ in rutile (Taylor-Jones and Powell, 2015), which would play a role in a further reaction. Excess Si, possibly derived from the fragmentation and dissolution of garnet and sillimanite, would form quartz veins and react with Zr to form zircon coronae. As for the apatite formation, P, F, Cl and OH could be derived either from decomposed biotite or were delivered by the water-rich fluid as components of a water brine from, for example, dissolution of monazite. However, the occurrence of apatite needles inside altered biotite grains points to genetic relationships between these two minerals.

Further cooling caused nonequilibrium recrystallization of rutile. Ewing et al. (2013) described partial replacement of rutile by other phases, characteristic for all granulitic metapelites from the IVZ. In our sample, we observe recrystallization of rutile with the formation of fine rutile–quartz intergrowths and thin zircon coronae around them. This took place during later stages of the rock evolution, when the temperature decreased and caused the Zr solubility in rutile to decrease. Therefore, we suggest the following stylized reaction (R1):



in which Rut_{Zr-Hf} (= SiO₂-, Zr- and Hf-bearing rutile) resulted from decomposition of ilmenite and Rut₂ (= SiO₂-, Zr- and Hf- depleted rutile) forms intergrowth with quartz (Qtz) and zircon coronae (Zrn_{cor}). Quartz was exsolved from SiO₂-rich rutile and formed thin intergrowths with the newly crystallized, trace-element-depleted rutile (e.g., Figs. 3, 4c). Zircon coronae produced in this way are very thin, suggesting slow reaction rates and a limited solubility of Zr in rutile at a low temperature (Tomkins et al., 2007; Ewing et al., 2013, 2014; Pape et al., 2016).

5.2 Zircon textures

Zircon grains hosted by garnet, sillimanite and fine-grained phyllosilicate matrix (Fig. 2a–c accordingly) represent detri-

tal grains, enclosed within the main mineral phases during metamorphism. Zircon in garnet shows euhedral shapes and concentric growth zoning (Fig. 2a), indicating capture of detrital grains by metamorphic garnet. Zircon enclosed in sillimanite has detrital cores that are overgrown by metamorphic rims, which are in growth equilibrium with sillimanite (Figs. 2b, 5c). After the peak metamorphic conditions these detrital zircon grains seem to have been mostly inert and are therefore well-preserved. Zircon grains hosted by the fine-grained phyllosilicate matrix in hydration veins are the most deformed and fractured and show dissolved and/or corroded surfaces (Fig. 2c). These latter grains were probably exposed to the post-peak metamorphic fluids. The dissolved material from their surfaces might have been transported with a fluid and serve as a source for the zircon coronae precipitation. We suggest, however, that this was not the main source of Zr for coronae zircon and that coronae mainly precipitated from Fe–Ti oxides.

Vermicular-shaped aggregates of zircon (or thick coronae) and thin coronae (Figs. 3–5) have a different origin than the detrital grains (Fig. 2). It has been shown that zircon can grow from other Zr-bearing phases as a result of mineral reactions and as a mineral response to the changing conditions (Bingen et al., 2001; Söderlund et al., 2004; Austrheim et al., 2008; Ewing et al., 2013, 2014; Kohn et al., 2015; Pape et al., 2016). Metamorphic (coronae) zircon in granulite-facies rocks may not be a product of peak metamorphism, but precipitate during the retrograde evolution (Tomkins et al., 2007). Zircon coronae textures are the evidence of zircon formation due to breakdown of Zr-bearing Fe–Ti oxides (e.g., Davidson and van Breemen, 1988; Fraser et al., 2004; Degeling et al., 2001). However, taking into account complexity of the textures and the fact that they were formed in more than one stage, we do not entirely exclude the possibility of exsolution of zircon from, for example, Zr-bearing rutile or exsolution from ilmenite before its breakdown (e.g., Bingen et al., 2001; Tomkins et al., 2007; Ewing et al., 2013; Pape et al., 2016). The possibility of the two-stage exsolution of Zr from Fe–Ti oxides was suggested by Ewing et al. (2013). Our textural observations are consistent with this idea. The sketch in Fig. 6 shows stages (1) and (2) of zircon coronae formation:

1. After the peak metamorphic conditions, ilmenite was the main host phase for Zr, together with the primary detrital zircon (Bingen et al., 2001). At the initial cooling stage it partially decomposed to rutile (Ewing et al., 2013). The expelled Zr was not entirely incorporated into the growing rutile and precipitated as new zircon (Fig. 6). Formation of zircon vermicular aggregates preceded brittle and ductile deformation of the rock. Vermicular grains in 3-D volume represent curved envelope-type aggregates (Fig. 4b–e), thus resembling coronae in shape (e.g., Bingen et al., 2001). However, they are thicker than what was previously observed

for zircon. Therefore, we interpret vermicular grains as evolved coronae. The thickness of these coronae should be controlled by reaction and cooling rates (Kohn et al., 2015). At comparatively high temperatures and slow reaction rates, zircon coronae grew thick, and formed lamellae-like intergrowths with the newly forming rutile (e.g., Fig. 4a–b). Formation of similar exsolution lamellae was described for many metamorphic minerals (e.g., Zhang and Liou, 2000).

2. In contrast with the thick coronae, formation of thin zircon coronae during Reaction (R1) occurred at lower temperatures, simultaneous with or soon after fracturing, as some of these coronae fill fractures (Figs. 3a, e, 5c). Fracture filling also indicates local Zr mobility, aided by fluid. At lower temperatures rutile recrystallizes and progressively incorporates less Zr (Ewing et al., 2013) than the high-temperature rutile, according to Zr-in-rutile thermometer models (Watson et al., 2006; Ferry and Watson, 2007). Thus, the excess Zr in the cooling system should be hosted by other Zr-bearing phases, most commonly by zircon (e.g., Pape et al., 2016). Crystallization of thin ($\leq 1 \mu\text{m}$) zircon coronae and thin needle-shaped 1–3 μm long rutile grains indicates rapid cooling resulting in non-equilibrium recrystallization of Zr-bearing rutile, when Zr and Hf were expelled from the host grain (Ewing et al., 2013). This occurred after the initial cooling during the exhumation stage (Ewing et al., 2014; Kohn et al., 2015). The rutile grains that did not recrystallize usually occur in intimate contact with the rutile–quartz intergrowths and are separated from them by zircon coronae (e.g., Fig. 4c). Thus, it is possible to indicate the direction of the recrystallization front. Earlier zircon grains serve as the nucleation spots for thin zircon coronae, which is similar to the low-temperature textures described by Rasmussen (2005).

Zircon coronae in our sample are different from those described in Bingen et al. (2001), Charlier et al. (2007), and Morisset and Scoates (2008), who only observed coronae at the boundary of the (former) ilmenite grains. The described textures are also different from the coronae reported by Austrheim et al. (2008) and Pape et al. (2016), in which zircon forms continuous chains or closed contours of small grains. However, zircon coronae in all cases (described in earlier literature and here) represent 3-D shells and/or envelopes around the reacting grains (Bingen et al., 2001). Textures, indicating reaction fronts of rutile recrystallization, have not been found by Pape et al. (2016), even though these authors searched these features. In contrast, in our sample we observe former reaction fronts formed by tangled and split zircon coronae within recrystallized rutile aggregates. Split coronae may show different reaction fronts converging to one point (Figs. 3e, 4e). The reaction fronts moved from rutile–quartz intergrowths towards unreacted rutile, forming rutile–quartz

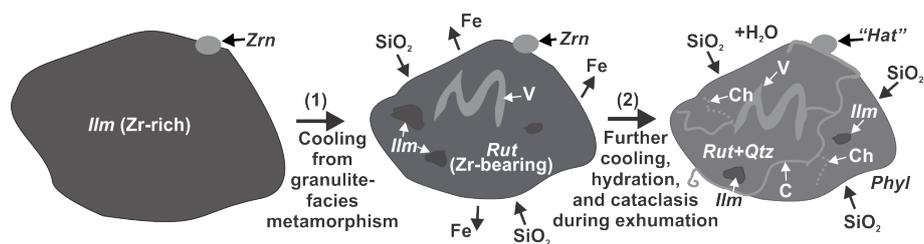


Figure 6. Sketch of the formation stages of zircon coronae. Post-peak metamorphic Zr-rich ilmenite fills the pocket between peak metamorphic minerals and has lobate boundaries. During the initial retrograde cooling (1) it decomposes to Zr- and SiO_2 -bearing rutile and vermicular-shaped zircon aggregates. The system loses Fe and requires SiO_2 from the surrounding phases; the volume of Fe–Ti oxide decreases. Further cooling, hydration and cataclasis during exhumation (2) results in recrystallization of Zr-bearing rutile to rutile–quartz intergrowths with precipitation of thin zircon coronae. At this point the reaction requires SiO_2 and aqueous fluid from the surroundings. Abbreviations as in Figs. 1, 3 and 5.

embayments in the latter, rimmed by zircon coronae (e.g., Fig. 4c, arrows show the directions of the reaction front). The chains of small zircon grains are effectively the same as zircon coronae and are similar to those described in Austrheim et al. (2008). The hat-shaped zircon grains are formed by coronae that are connected to the larger zircon grains (Fig. 5a) and thus represent aggregates formed by different zircon generations.

Not all rutile aggregates in our sample are associated with zircon coronae. Similarly, the diversity in appearance of rutile grains from the same sample was described by Pape et al. (2016) for IVZ metapelites. This can be due to (a) thin section cut that does not reveal associated coronae or (b) only local recrystallization of rutile (e.g., due to locally elevated strain or inhomogeneous distribution of fluid), so that the rest of the rutile still contains a significant amount of Zr. In the case of (b), Zr-in-rutile thermobarometry can be applied to the Zr-enriched rutile to estimate the temperature of ilmenite decomposition and coeval formation of vermicular zircon (e.g., Ewing et al., 2013).

6 Conclusions and implications

In our study, we demonstrate that zircon coronae can form within and around Fe–Ti oxides in metapelites during cooling and hydration after peak granulite-facies metamorphism. Zircon formed as a result of breakdown (exsolution) of ilmenite and rutile. Formation of zircon coronae occurred in two distinct stages and resulted in (1) thick (5–20 μm) vermicular-shaped grains presumably formed during breakdown of Zr-bearing ilmenite to Zr-bearing rutile, and in (2) thin ($\leq 1 \mu\text{m}$) corona aggregates and submicron-grain chains formed due to low-temperature recrystallization of Zr-bearing rutile (Fig. 6). Two zircon-forming episodes were separated in time and represent two evolution stages of the sampled rock and could therefore be connected with the evolution of the Ivrea–Verbano Zone on a larger scale.

We report a new textural relationship between zircon and host rutile grains, as only exsolution needles of zircon in rutile and small zircon grains framing rutile were described in metapelites before (e.g., Ewing et al., 2013; Pape et al., 2016). We describe zircon coronae in metasedimentary rocks, in contrast with the previous authors, who reported similar textures in metaigneous rocks (e.g., Bingen et al., 2001; Austrheim et al., 2008 and references therein).

The detailed study of zircon corona textures can have a significant influence on the trace element balance calculations for the bulk rock, provides a tool for the reconstruction of metamorphic mineral–fluid reactions and helps derive the direction of rutile recrystallization reaction fronts. Moreover, precipitated zircon can potentially be used in geochronology for in situ dating of metamorphic evolution stages and may yield the isotopic age of metamorphic reactions (e.g., Charlier et al., 2007; Ewing et al., 2013). The trace elements in zircon can be measured to fingerprint different fluid infiltration–recrystallization events. They can be used in thermobarometry for estimating the P – T conditions of the ilmenite breakdown and formation of Zr-bearing rutile and the P – T conditions of the Zr exsolution from rutile. For the latter, Zr-in-rutile, Ti-in-zircon and Si-in-rutile thermometers can be applied (e.g., Ewing et al., 2013; Pape et al., 2016).

Data availability. There are no datasets associated with the manuscript other than those included in the paper and the Supplement.

The Supplement related to this article is available online at <https://doi.org/10.5194/se-8-789-2017-supplement>.

Author contributions. EK and UK were responsible for sampling. EK performed laboratory work, SEM and EMPA analysis, data reduction and analysis, and drafted the paper. HA and UK conceptu-

alized the study, oversaw the progression of the work and advised on interpretation.

Competing interests. The authors declare that they have no conflict of interest.

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